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WO 02/17973 A2

(54) Title: **PREPARING STERILE ARTICLES FROM CERTAIN POLYMERS**

(57) Abstract: A method of preparing a sterile article is disclosed. An article is prepared from a polymer into which is incorporated a stabilizer having the general formula: where each A is independently selected from E, OR, SR, and CO-R, and can join the aromatic ring to form an additional ring, E is R or R(OCH₂R"CH)_n, G is E, CO-R, -C(E)H-O-R, C(OE)(H)-O-R, -C(E)(R")R"-OR, -C(E)(R")-R"-CO-OR, C(OE)(R")-E"-COOR, -C(OE)R"-O-R, -C(OE)HR, or -C(OE)R₂, J is A, each R is independently selected from H, R', R"OR', R"COOR', R"C(OR")R", or R"C(OR")R"OR", each R' is independently selected from alkyl from C₁ to C₂₄, aryl from C₆ to C₂₄, alkaryl from C₇ to C₂₄, and aralkyl from C₇ to C₂₄, R" is H or R', R" is alkylene from C₁ to C₂₄, arylene from C₆ to C₂₄, alkarylene from C₇ to C₂₄, or aralkylene from C₇ to C₂₄, m is 1 to 7, and n is 1 to 20. The article is exposed to gamma radiation; it yellows less after irradiation than it otherwise would.

PREPARING STERILE ARTICLES
FROM CERTAIN POLYMERS

Background of the Invention

This invention relates to the preparation of sterile articles from certain polymers. In particular, it relates to making the articles from poly(vinylchloride) (PVC) that contains a benzyl alcohol, cinnamyl alcohol, or geraniol-based stabilizer, then subjecting the article to ionizing radiation, particularly gamma radiation.

Various polymers, including PVC, polyethylene, and polypropylene, are used to make medical devices and as packaging for food. Sterilization is required and can be accomplished by exposing the devices or packages to gamma radiation. However, the gamma radiation can degrade the polymers, making them unsuitable or less acceptable for certain applications. In the case of PVC, the yellowing of sterilized articles is the most notable physical change as the result of such degradation.

Benzyl alcohol, geraniol, and some of the related ethers were reported to enhance or help enhance the thermal stability of PVC. (L. L. Wood GB 1151108, CAN 71:92299; J. D. Collins, H. Coates, I. H. Siddiqui US 3,845,017; JP 55069609 CAN 93:151163). Benzyl alcohol and cinnamyl alcohol have also been used as solvents for biocides in PVC. (EP 168949 CAN 104:187573)

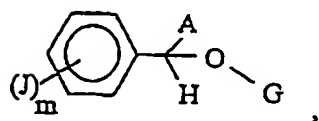
Summary of the Invention

I have discovered that when certain compounds based on benzyl alcohol, cinnamyl alcohol, or geraniol are incorporated into certain polymers and the polymers are made into articles and are sterilized with gamma radiation, substantially less yellowing occurs.

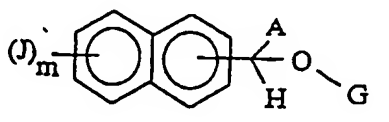
Most of the stabilizers used in this invention are FDA approved additives for food use and could be used for food and medical applications.

Description of the Preferred Embodiments

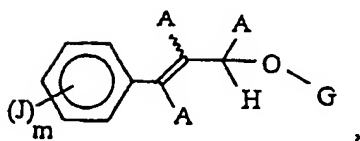
The stabilizers used in this invention have the general formula:



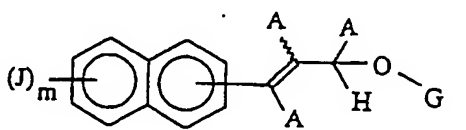
(A)



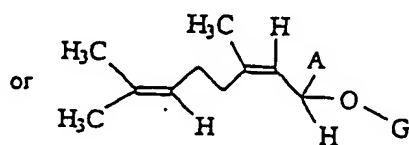
(B)



(C)



(D)



(E)

where each A is independently selected from E, OR, SR, and CO-R, and can join the aromatic ring to form an additional ring (e.g., α -hydroxyxanthene), E is R or $R(OCH_2R'CH)_n$, G is E, CO-R, $-C(E)H-O-R$, $C(OE)(H)-O-R$, $-C(E)(R')R''-OR$, $-C(E)(R')-R''-CO-OR$, $C(OE)(R')-E''-COOR$, $-C(OE)R''-O-R$, $-C(OE)HR$, or $-C(OE)R_2$, J is A, each R is independently selected from H, R', $R''OR'$, $R''COOR'$, $R''C(OR'')R''$, or $R''C(OR'')R''OR''$, each R' is independently selected from alkyl from C_1 to C_{24} , aryl from C_6 to C_{24} , alkaryl from C_7 to C_{24} , and aralkyl from C_7 to C_{24} , R'' is H or R', R''' is alkylene from C_1 to C_{24} , arylene from C_6 to C_{24} , alkarylene from C_7 to C_{24} , or aralkylene from C_7 to C_{24} , m is 1 to 7, and n is 1 to 20. Preferably, A is H and J is R, OR, or SR as those stabilizers are easier to make and more effective. Also, R is preferably alkyl from C_7 to C_{24} , R' is preferably alkyl from C_1 to C_{12} , R'' is preferably alkyl from C_1 to C_6 , R''' is preferably alkyl from C_1 to C_6 , G is preferably H or benzyl, and m is preferably 1 for the same reasons. The preferred stabilizers are general formulas (A) and (C) as they are easier to make and less expensive. Example of stabilizers of formula (A) include 4-benzyloxybenzyl alcohol, benzhydrol, 9,10-dihydro-9-anthracenol, 9-hydroxyxanthene, 9H-thioxanthen-9-ol, 1,2,3,4-tetrahydro-1-naphthalenol, benzyl ether, diveratryl ether, benzyl methoxyethyl acetal, (1R)-(-)-nopol benzyl ether, 1,3-dibenzyloxy-2-propanol, benzyl benzoate, benzyl acetoacetate, benzyl phenylacetate, dibenzyl adipate, dibenzyl malonate, dibenzyl sebacate, benzyl butyl phthalate, poly(vinyl benzyl ether), poly(benzyl methacrylate), poly(methyl 4-benzyloxy-phenylsiloxane); Example of stabilizers of formula (B) include 1-naphthalenemethanol, 2-naphthalenemethanol, 1-

[(phenylmethoxy)methyl]naphthalene, 2-[(phenylmethoxy)-methyl]naphthalene, 1,1'-
[oxybis(methylene)]-bis-naphthalene, 2,2'-[oxybis(methylene)]-bisnaphthalene, 1-
naphthylmethyl acetate; Example of stabilizers of formula (C) include cinnamyl
alcohol, -amyl cinnamyl alcohol, trans-2-methyl-3-phenyl-2-propen-1-ol, cinnamyl
5 ether, benzyl cinnamyl ether, cinnamyl acetate, trans-cinnamyl butyrate, cinnamyl
benzoate, cinnamyl phenylacetate, cinnamyl isovalerate, cinnamyl acid cinnamyl
ester, poly(vinyl cinnamyl ether), poly(cinnamyl methacrylate), poly(methyl 4-
cinnamyloxy-phenylsiloxane); Example of stabilizers of formula (D) include 3-(1-
naphthyl)-2-propen-1-ol, 3-(2-naphthyl)-2-propen-1-ol, 3-(1-naphthalenyl)-2-propen-
10 1-ol acetate, 3-(2-naphthalenyl)-2-propen-1-ol acetate; Example of stabilizers of
formula (E) include geraniol, benzyl geranyl ether, geranyl hexanoate, geranyl
phenylacetate, poly(vinyl geranyl ether), poly(geranyl methacrylate), poly(methyl 4-
geranyloxy-phenylsiloxane).

Some of the stabilizers are commercially available as food additives. Those
15 that are not commercially available can be made by techniques known in the art. For
example, cinnamyl benzyl ether can be made by the benzylation of cinnamyl alcohol
with benzyl chloride or benzyl bromide in the presence of a base.

The stabilizers of this invention are effective in reducing the yellowing of
articles made from polymers such as PVC, polyvinylidene chloride, polyethylene, and
20 polypropylene after exposure to ionizing radiation. The polymer can be made into
any type of article, including medical devices, food packaging, and radiation
protection devices.

The polymer can be stabilized by the addition of about 0.005 to about 70 phr (parts by weight per 100 part by weight of the polymer) of the stabilizer to the polymer. Less stabilizer is less effective and more stabilizer offers little additional benefit. The preferred amount of stabilizer is about 0.2 to about 20 phr. The stabilizer can be added to a polymer in a variety of ways, such as mixing the reactants at the beginning or during polymerization. The stabilizer is preferably added after at least 70 wt% of the monomer has polymerized. The stabilizer can be added as a solid or with a solvent as a slurry or a solution. Common organic solvents such as N-methylpyrrolidone, diglyme, acetamide, acetone, methanol, ethanol, isopropanol, dimethylsulfoxide, or dimethylformamide can be used; water can also be used. Water miscible solvents, such as acetone, tetrahydrofuran, and methanol, are preferred. If the stabilizer is a solid, it is preferable to add it in a solvent as that achieves a more uniform distribution of the stabilizer in the polymer. The stabilizer can also be added to the polymer along with or as a shortstop, or during the drying or compounding of the polymer. Various methods can be used for compounding, including milling, dry mixing, and extrusion.

The following examples further illustrate this invention:

EXAMPLES 1 to 125

To a mixture of 150.00 g PVC (sold by Occidental Chemical Corporation as "Oxy 240" or sold by Oxy Vinyls as "OV 30"), 0.30 g stearic acid (used as a lubricant), 0.23 g of a zinc and calcium mixed salts of mixed fatty acids (used as a

heat stabilizer; sold by Witco as "Mark 152 S"), 97.50g (less the amount of gamma ray stabilizer used) of dioctyl phthalate ("DOP," used as a plasticizer and to increase flexibility), 15.00 g epoxidized soy bean oil (used as an HCl scavenger to reduce degradation; sold by Witco as "Drapex 6.8"), was added a gamma ray stabilizer. The mixture was thoroughly blended and hot milled at 300°F (182°C) for 5 minutes. The resulting PVC sheet was cut and pressed into a 4"x3"x1/4" (10x8x0.6 cm) plaque at 330°F (182°C). The plaque was divided into two smaller pieces. One was saved for comparison and one was subjected to γ radiation at a dose of 50 kGy. The irradiated piece was again divided into two pieces and one of these pieces was oven aged at 50°C for 48 hours. All of the samples were measured for yellowness index with a Macbeth 2020 Plus Color Eye Spectrometer, as described by the Hunter equations (see "The Measurement of Appearance" by Richard S. Hunter, John Wiley & Sons, New York, 1975). The following table gives the stabilizers used and the results.

Example	Stabilizer	Amount (g)	Yellowness		
			Initial	After Radiation	After Aging
Control	none	none	19.1	55.4	79.9
1	Benzyl ether	0.5	16.2	42.9	65.1
2	"	1.0	17.0	37.0	57.2
Control	none	none	21.1	57.2	77.2
3	Benzyl ether	2.01	32.8	33.1	51.9
4	"	5.02	19.4	26.9	42.4
5	"	10.02	17.6	22.3	36.5
6	"	15.04	15.6	19.7	32.2
7	"	20.03	16.1	18.8	31
8	"	30.02	15.4	17.3	27.5
Control	none	none	17.1	52.3	84.4
9	Benzyl methoxyethyl acetal	0.5	17.9	42.1	66.1
10	"	1.02	16.8	35.1	58.9
11	"	1.5	16.5	34.6	54.5

12	"	2.02	15.4	29.7	50.5
13	"	2.52	17.2	27.8	47.8
14	"	3.02	16.5	27.5	45.2
Control	none	none	19.3	55.7	75.4
15	Diveratryl ether	1.30	19.2	39.8	59.7
16	"	3.50	20.6	33.2	50.8
17	Benzyl cinnamyl ether	2.00	18.9	35.8	47.9
18	"	4.00	17.1	30.5	37.8
19	"	6.00	16.1	27.4	36.4
Control	none	none	19.1	55.4	79.9
20	(1R)-(-)-Nopol benzyl ether	6.46	19.3	38.6	54.7
21	"	12.92	16.7	34.8	47.0
Control	none	none	19.1	55.4	79.9
22	1,3-Dibenzyl oxy-2-propanol	0.51	17.5	49.3	70.7
23	"	1.02	16.6	44.4	64.3
24	"	1.54	17.5	42.8	62.4
25	"	2.03	17.4	39.8	60.3
Control	none	none	18.1	50.2	75.5
26	4-Benzyloxybenzyl alcohol	2.40	18.6	28.4	42.1
27	"	4.80	18.1	25.5	34.7
Control	none	none	19.1	55.4	79.9
28	9-Hydroxyxanthene	0.50	17.1	62.9	84.2
29	"	1.00	18.4	51.4	72.7
30	"	1.52	17.3	46.7	68.4
31	"	2.17	17.7	42.2	63.7
Control	none	none	21.1	57.2	77.2
32	Benzyl benzoate	2.01	18.6	45.7	65.3
33	"	5.1	19.0	39.7	58.5
34	"	10.01	18.7	34.3	51.8
35	"	15.03	18.1	29.9	47.2
36	"	20.19	18.9	31.7	50.5
37	"	30.08	19.1	28.3	43.8
Control	none	none	19.1	55.4	79.9
38	benzyl acetoacetate	4.88	17.9	34.1	46.1
39	"	9.72	23.0	31.0	42.4
Control	none	none	21.1	57.2	77.2
40	Dibenzyl malonate	2.70	16.9	29.4	45.3
41	"	6.43	17.8	30.8	48.6
Control	none	none	19.3	55.7	75.4
42	Dibenzyl sebacate	3.05	18.8	39.9	59.4
43	"	6.06	18.5	34.2	52.1
44	"	9.08	17.7	33.1	48.8
Control	none	none	19.3	55.7	75.4

45	Dibenzyl adipate	3.00	18.0	40.0	58.3
46	"	6.01	17.9	35.3	52.8
47	"	9.02	21.1	32.0	47.8
Control	none	none	19.3	55.7	75.4
48	Benzyl butyl phthalate	10.03	18.0	40.2	60.7
49	"	20.03	18.1	35.8	53.7
50	"	30.05	18.8	32.5	48.3
51	"	40.04	18.5	32.0	46.6
52	"	50.02	18.7	30.6	45.0
53	"	60.06	20.1	29.2	43.6
Control	none	none	17.1	52.3	84.4
54	Dibenzyl phthalate	1.50	17.7	54.3	80.4
55	"	4.29	15.0	47.7	70.9
Control	none	none	17.1	52.3	84.4
56	Cinnamyl alcohol	0.54	17.0	37.6	62.5
57	"	1.11	14.7	27.6	48.4
58	"	1.53	17.2	27.5	42.8
59	"	2.03	15.5	21.9	35.9
60	"	2.6	16.9	23.6	34.7
61	"	3.09	14.6	20.0	31.5
Control	none	none	17.1	52.3	84.4
62	Alpha-amyl cinnamyl alcohol	0.51	18.7	56.9	77.3
63	"	1.01	18.0	57.1	77.1
64	"	1.5	17.6	57.6	74.2
65	"	2.01	18.9	59.0	74.7
66	"	2.5	19.3	57.7	72.8
67	"	3.02	18.9	57.2	70.0
Control	none	none	16.9	61.7	84.1
68	Trans-2-methyl-3-phenyl-2-propen-1-ol	1.01	17.3	34.8	55.4
69	"	2.00	14.4	27.8	44.0
70	"	3.00	16.3	24.2	37.8
Control	none	none	19.3	55.7	75.4
71	Cinnamyl ether	1.0205	19.7	44.1	55.9
72	"	2.0412	20.7	37.1	46.4
73	"	3.0637	18.8	33.1	41.9
74	"	5.1007	18.8	29.0	36.3
75	"	8.1642	18.2	29.2	37.3
Control	none	none	19.3	55.7	75.4
76	Cinnamyl acetate	1.39	19.7	39.6	51.1
77	"	3.30	19.3	36.8	51.6
Control	none	none	16.6	49.1	52.5
78	Trans-cinnamyl butyrate	1.00	18.2	40.2	46.5

79	"				
80	"	2.02	17.5	35.5	38.7
81	"	3.00	17.4	37.3	37.5
82	"	4.01	17.0	33.9	38.1
83	"	5.01	16.5	33.2	36.3
Control	none	6.02	16.0	32.8	34.3
84	Cinnamyl benzoate	none	19.3	61.6	79.0
85	"	3.03	18.8	52.2	64.4
86	"	6.00	18.5	44.0	56.5
Control	none	9.02	18.6	42.1	52.8
87	Cinnamyl phenylacetate	none	19.3	61.6	79.0
88	"	3.01	18.3	43.8	56.9
89	"	6.02	18.7	39.5	49.7
Control	none	9.04	18.7	36.4	46.8
90	Trans-cinnamyl isovalerate	none	19.3	61.6	79.0
100	"	3.01	18.6	47.6	58.5
101	"	6.01	18.5	42.7	52.6
Control	none	9.01	18.7	39.6	49.1
102	Cinnamic acid cinnamyl ester	none	19.3	55.7	75.4
103	"	3.03	18.7	42.6	52.0
104	"	6.04	19.3	38.3	45.9
Control	none	9.05	19.3	35.9	45.1
105	Geraniol	none	14.9	54.0	74.7
106	"	0.49	14.2	48.1	67.5
107	"	0.99	15.0	37.7	56.6
108	"	2.10	14.4	32.9	50.4
Control	none	2.99	15.1	31.2	45.1
109	Geranyl hexanoate	none	17.1	52.3	84.4
110	"	0.51	15.6	46.0	72.0
111	"	1.01	17.5	45.2	73.3
112	"	1.52	14.8	42.5	70.7
113	"	2.02	18.6	42.2	67.8
114	"	2.5	15.2	36.3	66.2
Control	none	3.03	17.3	37.9	67.8
115	Geranyl benzyl ether	none	16.6	49.1	51.5
116	"	1.03	15.5	36.2	44.7
117	"	2.01	15.0	33.3	35.1
118	"	3.00	14.2	26.7	33.2
119	"	4.00	14.2	26.1	33.2
120	"	5.01	14.7	23.4	35.1
Control	none	6.04	15.6	25.2	32.0
121	Geranyl phenylacetate	none	16.6	49.1	52.5
122	"	1.50	18.4	40.5	47.3
		3.00	15.7	35.1	40.2

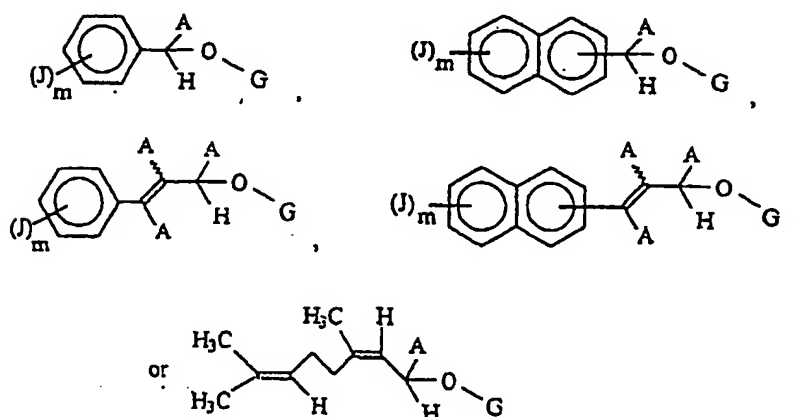
123	"	4.50	16.2	29.9	35.5
124	"	6.02	15.3	30.2	37.3
125	"	7.52	16.0	31.2	33.5

5

As can be observed from the above table, after γ radiation the polymers that contained a stabilizer had significantly less discoloration than the control samples. The table also shows that ethers and alcohols are better at reducing yellowing than esters and can be used in lower amounts.

CLAIMS:

1. A method of preparing a sterile article comprising
- (A) producing a polymer selected from the group consisting of poly(vinyl chloride), poly(vinylidene chloride), polyethylene, and polypropylene, that contains about 0.005 to about 70 phr of a stabiliser having the general formula:



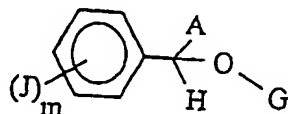
where each A is independently selected from E, OR, SR and CO-R, and can join the aromatic ring to form an additional ring, E is R or R(OCH₂R''CH)_n, G is E, CO-R, -C(E)H-O-R, C(OE)(H)-O-R, -C(E)R''R'''-OR, -C(E)(R'')-R'''-CO-OR, C(OE)(R'')-E'''-COOR, -C(OE)R'''-O-R, -C(OE)HR, or -C(OE)R₂, J is A, each R is independently selected from H, R', R'''OR', R'''COOR', R'''C(OR'')R'', or R'''C(OR'')R'''OR'', each R' is independently

selected from alkyl from C₁ to C₂₄, alkaryl from C₇ to C₂₄, and aralkyl from C₇ to C₂₄, R'' is H or R', R'' is alkylene from C₁ to C₂₄, arylene from C₆ to C₂₄, alkarylene from C₇ to C₂₄, or aralkylene from C₇ to C₂₄, m is 1 to 7, and n is 1 to 20;

(B) making said article from said polymer, and

(C) sterilising said article with ionising radiation.

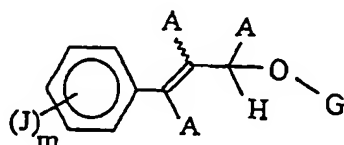
2. A method according to Claim 1 wherein said polymer comprises poly(vinylchloride).
3. A method according to Claim 1 or 2 wherein A is H.
4. A method according to Claim 1, 2 or 3 wherein G is H or benzyl.
5. A method according to any one of Claims 1 to 4 wherein J is R, OR, or SR.
6. A method according to any one of claims 1 to 5 wherein R is alkyl from C₇ to C₂₄.
7. A method according to any one of Claims 1 to 6 wherein said stabiliser has the general formula



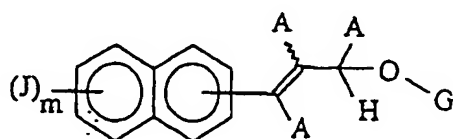
8. A method according to any one of Claims 1 to 6 wherein said stabiliser has the general formula



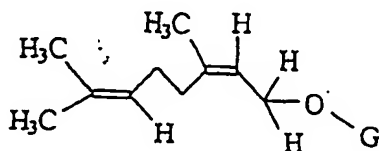
9. A method according to any one of Claims 1 to 6 wherein said stabiliser has the general formula



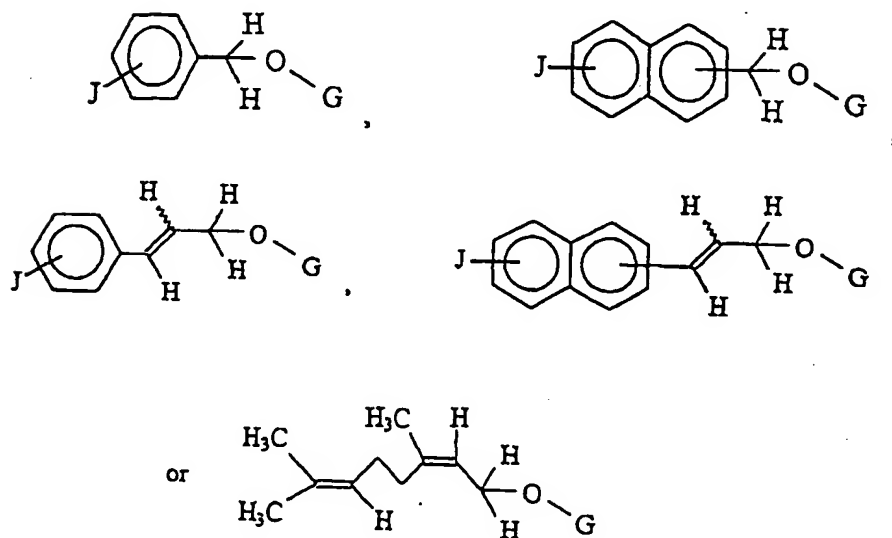
10. A method according to any one of Claims 1 to 6 wherein said stabiliser has the general formula



11. A method according to any one of Claims 1 to 6 wherein said stabiliser has the general formula



12. A sterile article prepared according to the method of any one of Claims 1 to 11.
13. A method preparing a sterile article comprising
- (A) making poly(vinyl chloride) that contains about 0.2 to about 20 phr of a stabiliser having the general formula

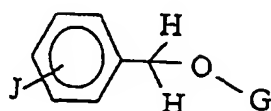


where G is H or benzyl and J is R, OR, or SR, and R is alkyl from C₇ to C₂₄.

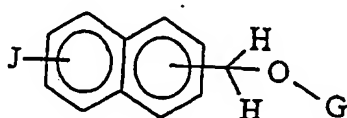
- (B) making said article from said poly(vinyl chloride), and

(C) sterilising said article with gamma radiation.

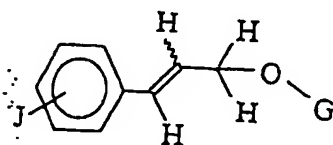
14. A method according to Claim 13 wherein said stabiliser has the formula



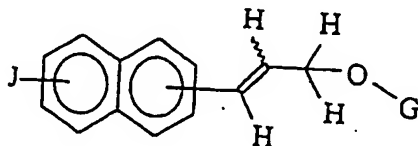
15. A method according to Claim 13 wherein said stabiliser has the general formula



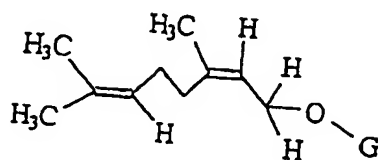
16. A method according to Claim 13 wherein said stabiliser has the general formula



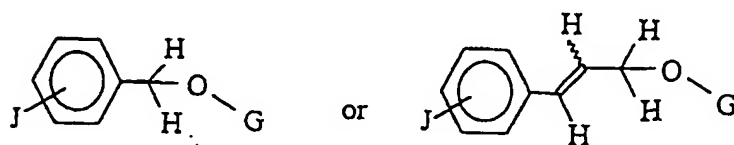
17. A method according to Claim 13 wherein said stabiliser has the general formula



18. A method according to Claim 13 wherein said stabiliser has the general formula



19. A sterile medical article prepared according to the method of any one of Claims 13 to 18.
20. A method of preparing a sterile medical article comprising
- (A) making poly(vinyl chloride) that contains about 0.2 to about 20 wt % of a stabiliser having the general formula

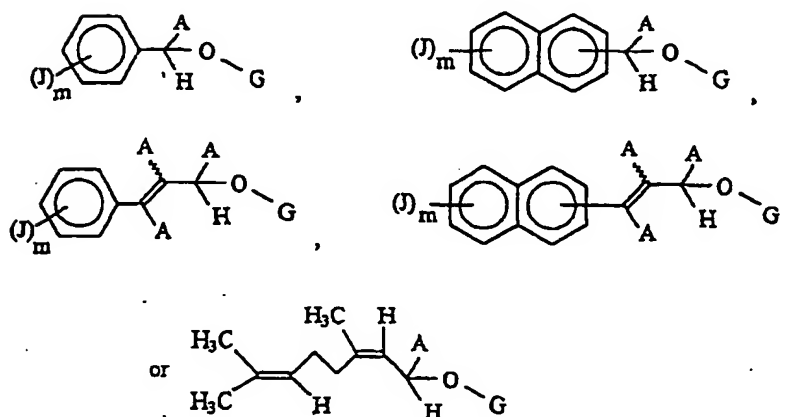


where G is H or benzyl and each R is independently selected from alkyl from C₈ to C₁₂,

- (B) making said medical article from said poly(vinyl chloride), and
- (C) sterilising said medical article with gamma radiation.

21. A resin comprising

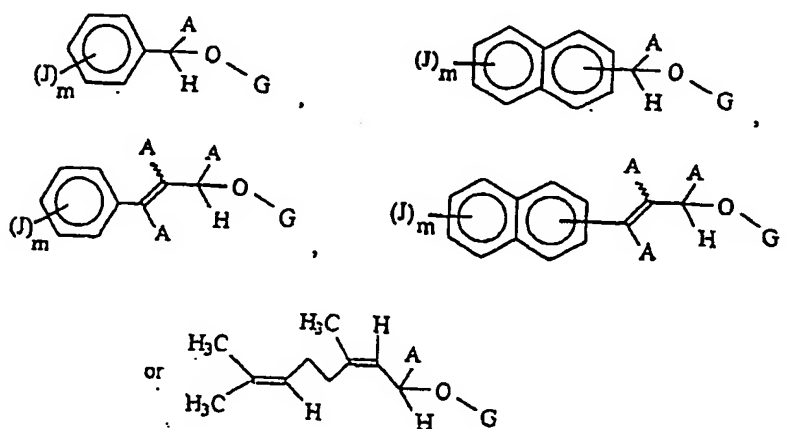
- (A) a polymer selected from the group consisting of poly(vinyl chloride), poly(vinylidene chloride), polyethylene and polypropylene that contains about 0.005 to about 70 phr of a stabiliser having the general formula



where each A is independently selected from E, OR, SR, and CO-R, and can join the aromatic ring to form an additional ring, E is R or $R(OCH_2R''CH)_n$, G is E, CO-R, $-C(E)H-O-R$, $C(OE)(H)-O-R$, $-C(E)(R'')R'''-OR$, $-C(E)(R'')-R'''-CO-OR$, $C(OE)(R'')-E'''-COOR$, $-C(OE)R'''-O-R$, $-C(OE)HR$, or $-C(OE)R_2$, J is A, each R is independently selected from H, R' , $R'''OR'$, $R'''COOR'$, $R'''C(OR'')R''$, or $R'''C(OR'')R'''OR''$, each R' is independently selected from alkyl from C_1 to C_{24} , aryl from C_6 to C_{24} , and alkaryl from C_7 to C_{24} , R'' is H or R' , R''' is alkylene from C_1 to

C_{24} , arylene from C_6 to C_{24} , alkarylene from C_7 to C_{24} , or aralkylene from C_7 to C_{24} , m is 1 to 7, and n is 1 to 20.

22. A stabiliser for use in stabilising a polymer, the stabiliser having the general formula



where each A is independently selected from E, OR, SR and CO-R, and can join the aromatic ring to form an additional ring, E is R or $R(OCH_2R''CH)_n$, G is E, CO-R, $-C(E)H-O-R$, $C(OE)(H)-O-R$, $-C(E)R''R'''-OR$, $-C(E)(R'')-CO-OR$, $C(OE)(R'')-E'''-COOR$, $-C(OE)R'''-O-R$, $-C(OE)HR$, or $-C(OE)R_2$, J is A, each R is independently selected from H, R' , $R'''OR'$, $R'''COOR'$, $R'''C(OR'')R''$, or $R'''C(OR'')R'''OR''$, each R' is independently selected from alkyl from C_1 to C_{24} aryl from C_6 to C_{24} , and aralkyl

from C₇ to C₂₄, R'' is H or R', R'' is alkylene from C₁ to C₂₄,
arylene from C₆ to C₂₄, alkarylene from C₇ to C₂₄ or aralkylene
from C₇ to C₂₄, m is 1 to 7, and n is 1 to 20;

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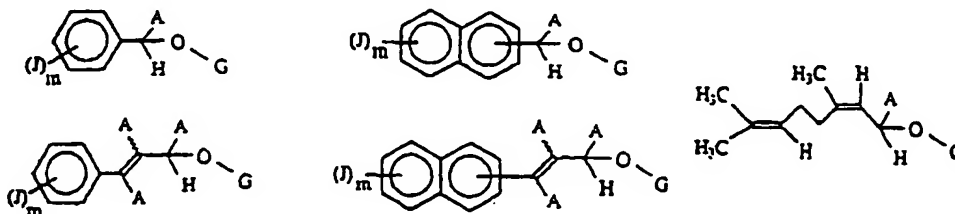
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(54) Title: PREPARING STERILE ARTICLES FROM CERTAIN POLYMERS



(57) Abstract: A method of preparing a sterile article is disclosed. An article is prepared from a polymer into which is incorporated a stabilizer having the general formula: where each A is independently selected from E, OR, SR, and CO-R, and can join the aromatic ring to form an additional ring, E is R or R(OCH₂R'^{''}CH)_n, G is E, CO-R, -C(E)H-O-R, C(OE)(H)-O-R, -C(E)(R'^{''})R^{'''}-OR, -C(E)(R'^{''})-R^{'''}-CO-OR, C(OE)(R'^{''})-E^{'''}-COOR, -C(OE)R^{'''}-O-R, -C(OE)HR, or -C(OE)R₂, J is A, each R is independently selected from H, R', R''OR', R''COOR', R''C(OR')R'', or R''C(OR')R''OR'', each R' is independently selected from alkyl from C₁ to C₂₄, aryl from C₆ to C₂₄, alkaryl from C₇ to C₂₄, and aralkyl from C₇ to C₂₄, R'' is H or R', R''' is alkylene from C₁ to C₂₄, arylenylene from C₆ to C₂₄, alkarylene from C₇ to C₂₄, or aralkylene from C₇ to C₂₄, m is 1 to 7, and n is 1 to 20. The article is exposed to gamma radiation; it yellows less after irradiation than it otherwise would.

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A. CLASSIFICATION OF SUBJECT MATTER

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According to International Patent Classification (IPC) or to both national classification and IPC

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Minimum documentation searched (classification system followed by classification symbols)

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Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 1 004 621 A (TEIJIN LTD) 31 May 2000 (2000-05-31) page 3, line 13 - line 24; claims 1,4,8	1,7,22
A	EP 0 345 662 A (GOODRICH CO B F) 13 December 1989 (1989-12-13) the whole document	1,2,21
A	EP 0 384 110 A (MOBAY CORP) 29 August 1990 (1990-08-29)	1,22
X	examples	22
P,X	WO 00 56812 A (STEPHENS DINAH ;OCCIDENTAL CHEM CO (US)) 28 September 2000 (2000-09-28) the whole document	1,8,10, 13,22

☐ Further documents are listed in the continuation of box C.

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Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 1004621	A	31-05-2000	EP 1004621 A1	31-05-2000
			WO 9964506 A1	16-12-1999
EP 0345662	A	13-12-1989	AU 3584389 A	07-12-1989
			EP 0345662 A1	13-12-1989
			JP 2043245 A	13-02-1990
EP 0384110	A	29-08-1990	US 5006572 A	09-04-1991
			CA 2002410 A1	24-07-1990
			DE 69023937 D1	18-01-1996
			DE 69023937 T2	25-04-1996
			EP 0384110 A1	29-08-1990
			JP 2232258 A	14-09-1990
			JP 2654216 B2	17-09-1997
WO 0056812	A	28-09-2000	AU 2119900 A	09-10-2000
			EP 1169380 A1	09-01-2002
			WO 0056812 A1	28-09-2000
			US 2001007884 A1	12-07-2001